Research Article

Development and Evaluation of Floating Microspheres of cephalexin as Gastroretentive Dosage Form

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ABSTRACT

Floating microsphere of cephalexin were prepared and evaluated with an objective to prolong the gastric residence time of the drug which may result in improved bioavailability, reducing dose frequency, improving solubility of drug and to provide stability of drug which are degraded by the higher pH of the intestine. The floating microspheres were prepared using different polymers like ethylcellulose, HPMC- Ethyl cellulose, cellulose acetate and Eudragit S-100 by solvent evaporation/ diffusion method. The major advantage of preparation technique includes short processing time, lack of exposure of the ingredients to high temperature and high encapsulation efficiency. The floating microspheres were evaluated for micrometric properties, particle size, percent yield, drug polymer compatibility (IR study), scanning electron microscopy and in vitro drug release. The micrometric properties were found to be good and scanning electron microscopy shows that spherical microsphere with porous surface were formed. The practical yield was more than 70% with a particle size range of 128.23-389.72 m. the percent entrapment is about 65% and more in larger particle as compared to smaller particle. The percent buoyancy was more than 70% upto 12 hours. The particle size, percent yield, percent drug entrapment and percent buoyancy was increased significantly (p<0.05) with increase in . The *in vitro* release was slow and extended to more than 12 hours. *in vitro* release was significantly decreased (p<0.05) with increase in polymer concentration. Release obeys first order kinetics model and the drug release was diffusion controlled with Fickian or non Fickian transport depending upon the polymers. The prepared floating microspheres were stable. Hence it can be inferred that the floating microsphere of cephalexin may prolong drug release thereby improving bioavailability and enhance opportunity of absorption in stomach to prevent degradation of drug under alkaline pH.

Keywords: cephalexin, floating microspheres; buoyancy; in vitro release.

INTRODUCTION

Oral drug administration has been the predominant route for drug delivery due to the ease of administration, patient convenience and flexibility in formulations. However, it is a well accepted fact today that drug absorption throughout the GI tract is not uniform. Using currently utilized release technology, oral drug delivery for 12or even 24 hours is possible for many drugs that are absorbed uniformly from GI tract¹.

Cephalexin is a broad spectrum antibiotic used for the treatement of respiratory tract infection, skin and skin structure infection,genitourinary infections^{2.3}. In order to enhance bioavailability, the improvement of its solubility and dissolution characteristics is considered to be very effective. Floating drug delivery system is an effective technique which can easily enhance the bioavailability, reducing dose frequency, improving solubility of drug and to provide stability of drug which are degraded by

the higher pH of the intestine.

In the present study, cephalexin loaded microspheres were prepared using different polymers like ethylcellulose, Hydroxyl propyl methylcellulose, cellulose acetate and eudragit S-100, to prolong the drug release in upper gastrointestinal tract, where absorption of cephalexin is well confined.

MATERIALS AND METHODS Chemicals

Cephalexin was a gift sample from aurobindu pharmaceuticals,Hyderabad.Ethyle cellulose & HPMC purchase from SD Fine Che Ltd., Mumbai, Cellulose acetate from NR Chem, Mumbai, Eudragit S 100 fromEvonik industries. All the chemicals used in the present study were of AR Grade.

Preparation of cephalexin loaded floating microspheres^{4,5,6,7}

The floating microspheres loaded with

cephalexin were prepared by emulsion-solvent evaporation/ solvent diffusion method using different polymers as follows:

- 1. Using EC polymer 46: The drug and ethyl cellulose were mixed in dichloromethane at various ratio and triethyl citrate (10%) was added as plasticizer. The slurry was slowly introduced into the solution of polyvinyl alcohol (0.25%, 200ml). The resultant solution was stirred at 500 rpm for 1 hrs. This formed microsphere was filtered and dried at room temperature.
- Using HPMC and EC polymers⁵: Drug, HPMC and EC were dissolved in a mixture of ethanol and dichloromethane at room temperature. This was poured into 250 ml water containing 0.01%. Tween-80 maintained at 30-40°C and subsequently stirred for 45 min to allow the volatile solvent to evaporate. The microspheres formed were filtered, washed with water and dried overnight at room temperature.
- 3. Using cellulose acetate polymer⁶:Floating microsphere loaded with cephalexin were prepared by using solvent diffusion/ evaporation method. The polymer. cellulose acetate and drug was dissolved in ethyl acetate and acetone in different proportion. The resulting solution was added slowly to 150 ml of 0.1 M acidic solution containing 0.05% polyvinyl alcohol maintaining at 30- 400c. The emulsion was continuously stirred for 2 hours at the speed of 500 rpm. The floating microspheres collected were by decantation and then dried over night.
- 4. Using Eudragit S-100 polymer ⁷:Eudragit S-100 and a drug were dissolved in 8 ml ethanol, followed by the addition of 2 ml isopropanol and 5 ml dichloromethane. The polymer solution was slowly introduced into 1000 ml of 0.4% poly (vinyl alcohol) aqueous solution with stirring at 250 rpm using a mechanical stirrer equipped with a 3 blade propeller. The solution was stirred for 10 min and the microspheres were collected by filtration. The collected microspheres were dried for 12h at 50°C.

Characterization
microspheres,of
floatingThe prepared floatingmicrospheres were

The prepared floating microspheres were evaluated for drug carrier interaction using scanning electron microscopy (SEM) and FTIR (Perkin Elmer 1600 series) spectral studies. For SEM studies The sample was prepared by lightly sprinkling the powder on a double adhesive tape stuck to an aluminum stub. The stubs were then coated with gold to a thickness of about 300Aº under an argon atmosphere using a gold sputter module in a high vacuum evaporator. The coated samples then randomlv scanned were and photomicrographs were taken with a scanning electronmicroscope (Joel JSM-1600, Tokyo, Japan). : Fourier Transformation Infrared spectra of pure drug, polymer and drug loaded floating microspheres were obtained in KBr pellets at moderate scanning speed between 4000-200 cm-1 in a Perkin-Elmer Fourier Transformation Infrared spectroscope.

Evaluation of floating microspheres 1. Yield of microspheres¹⁵

The prepared microspheres were collected and weighed. The measured weight was divided by the total amount of all non-volatile components, which were used for the preparation of the microspheres.

Practical yield

Actual weight of product X 100

Weight of excipient & drug

2. Particle size⁹

The size was measured using an optical microscope under regular polarized light, and the mean particle size was calculated by measuring 100 particles with the help of a calibrated ocular micrometer.

3. Drug entrapment efficiency^{10,11}

Microspheres equivalent to 50 mg of the drug were taken for evaluation The amount of drug entrapped was estimated by crushing the microspheres and extracting with aliquots of 0.1N HCl and small quantity of methanol repeatedly. The extract was transferred to a 100 ml volumetric flask and the volume was made up using 0.1N HCl. The solution was filtered and the absorbance was measured after suitable dilution with 0.1N HCl spectrophotometrically (UV-1700, Shimadzu, Japan) at 262 nm against appropriate blank. The amount of drug entrapped in the microspheres was calculated by using the following formula:

Percentage drug entrapment = Amount of drug actually present x 100

Theoretically expected drug loaded

4. Micrometric properties¹²

The microspheres were characterized by their micrometric properties such as angle of repose, tapped density, compressibility index and true density.

a) Angle of repose

The angle of repose Θ of the microspheres, which measures the resistance to particle flow was calculated as:

tan⊖ =2 H

where 2H/D is the surface area of the freestanding height of the microsphere heap that is formed after making the microspheres flow from the glass funnel.

b) Tapped density and compressibility index

The tapping method was used to calculate tapped densities and compressibility index using tapped density:

Tapped density =

Mass of microspheres

Volume of microspheres after tapping

% compressibility index = $[1 - V/V_0] \times 100$

Here, V and V_0 are the volumes of the sample after and before the standard tapping respectively.

c) True density

Density of hollow microspheres is determined by immersing the microspheres in 0.02% tween-80 solution in a metal mesh basket. The microspheres that are sunk after this process are used for density measurement as carried out by the displacement method.

5. In vitro buoyancy⁸

An *in vitro* floating study was carried out usi ng 0.1NHCl containing 0.02% tween-80 as a dispersing medium. Microspheres were spread over the surface of 900 ml of dispersing medium at 37±0.5°C.A paddle rotating at 100 rpm agitated the medium. Each fraction of microspheres floating on the surface and those settled down were collectedat a predetermined time point. The collected samples were weighed after drying.

% floating microspheres = Weight of floating microspheres X 100

Initial weight of floating microspheres

6. *In vitro* dissolution studies^{4,13,14}

The in vitro release of drug from floating microspheres was carried out using paddle type Electrolab tablet dissolution tester USP XXIII. Drug loaded microspheres equivalent to 250 mg of drug was introduced into 900ml of the dissolution medium (0.1N HCl) maintained 37±0.5°C with paddle rotating at100 rpm Aliquots were withdrawn at regular interval and analyzed spectrophotometrically using Shimadzu-1700 UV-visible spectrophotometer. The dissolution studies were carried out in triplicate in 0.1N HCl for 12 hours. The volume of the dissolution medium was adjusted to 900 ml at every sampling time by replacing 5ml with same dissolution medium.

RESULTS AND DISCUSSION

All the floating Microspheres were prepared by solvent evaporation/ diffusion methods. Ethylcellulose floating microspheres were prepared by solvent evaporation method as shown in table-4 In this method, o/w emulsion was formed stabilize by ethylcellulose and after evaporation of organic solvent cephalexin get encapsulated by ethylcellulose. EC-HPMC floating microspheres were prepared by solvent evaporation method as shown in table-5. In this method, the emulsion was stabilized by tween-80 and the volatile solvent get evaporated leaving a solidified thin film at the interface between aqueous

phase and organic phase, where cephalexin gets encapsulated in the core-coat of polymers⁵. Cellulose acetate floating microspheres were prepared by solvent diffusion/ evaporation method as shown in table-6. When the polymer solution pour into aqueous phase, all the methanol or acetone will diffused along with small amount of ethyl acetate resulting in an interfacial polymer deposition leading to the formation of floating microspheres¹⁶. eudragit S-100 floating microspheres were prepared by solvent diffusion method as shown in table-1. In this method, the organic solvent ethanol gets diffused out of dispersed droplets with controlled rate in presence of dichloromethane or propanol. The acrylic polymer eudragit instantly solidified as thin film at the interface between aqueous and organic phase where cephalexin get encapsulated in core-coat of polymer76. The ratio of mixture of solvent was kept constant and the concentration of respective polymer was varied in all the preparation methods.

Physical characteristics of cephalexin floating microspheres were measured according to the

methods describe above. The results and listen in Table 2.

The invitro dissolution studies were performed by over a period of 12 hours and results are shown in figure 1&2. *in vitro* release was significantly decreased with increase in polymer concentration. Release obeys first order kinetics model and the drug release was diffusion controlled with Fickian or non Fickian transport depending upon the polymers. scanning electron microscopy shows that spherical microsphere with porous surface were formed

IR studies indicated that during the process of formulation no chemical reaction has been taken place. At the end of 12 hours all the formulation were found to be satisfactory in terms of excellent micrometric properties, percent yield ,percent drug entrapment efficiency *,in vitro* buoyancy and highest *in vitro* drug release of 89.75% in sustained manner with constant fashion over a extended period of time of12 hours. The results proved

that the cephalexin loaded floating microspheres Were better in releasing the drug in the sustained fasion for extended period of 12 hours.

CONCLUSION

The in vitro evaluation study of cephalexin loaded floating micro spheres was greatly improved when compared with those of conventional tablet dosageform. Among all formulations C1 with the drug: polymer ratio of (1:2) was found to be satisfactory in terms of excellent micrometric properties, percent yield (87.28%), percent drug entrapment efficiency (81.66%), *in vitro* buoyancy (88.59%), and highest *in vitro* drug release of 89.75% in sustained manner with constant fashion over a extended period of time of 12 hours.

The study reveals satisfactory results with a further scope of pharmacokinetic and pharmacodynamic evaluation.

centrose, EC and HPMC, centrose acetate & Eudragit 5-100								
FORMULATION CODE	Drug (mg)	EC(gm)	DICLORO METHANE (ml)/ Triethyl citrate (%)	CONTINUOUS PHASE				
E1	500	0.50	15 10	0.25% PVA aqeous solution				
E2	500	1	15 10					
E3	500	1.5	15 10					
	DRUG	HPMC/EC	ETHANOL:DOCHLOROMETHANE	CONTINUOUS PHASE				
HE1	500	0.5/ 1.0	1:2	250 ml water containing 0.01% w/v tween-80				
HE2	500	0.5/2.0	1:2					
HE3	500	0.5/3.0	1:2					
	DRUG	CELLULOSE ACETATE	ETHYLE ACETATE: ACETONE	CONTINUOUS PHASE				
C1	500	1	1:1					
C2	500	2	1:1	250 ml 0.1 M acidic				
C3	500	3	1:1	w/v polyvinyl alcohol				
	DRUG	Eudragit – S100	ETHANOL:DICHLOROMETHANE:PROPANOL	CONTINUOUS PHASE				
EU1	500	0.5	8:5:2	250 ml water containing				
EU2	500	1	8:5:2	0.4%				
EU3	500	1.5	8:5:2	w/v polyvinyl alcohol				

 Table1: Formulations of cephalexin loaded floating microspheres prepared using ethyl cellulose, EC and HPMC, cellulose acetate &Eudragit S-100

	U 1	5	U I	
	Mean particle size (mean±SD) (□m)	Practical yield (mean±SD) (%)	Drug entrapment efficiency (mean±SD) (%)	<i>In vitro</i> buoyancy (mean±SD) (%)
E1	316.15±22.25	72.56 0.67	78.33 🗆 🗆 1.52	81.42 🗆 🗆 2.45
E2	342.89±24.45	80.83 🗆 🗆 0.63	83.11 🗆 🗆 2.07	85.23 🗆 2.80
E3	371.28±18.14	83.28 0.57	86.31 🗆 🗆 1.05	88.33 🗆 3.14
HE1	262.50±23.81	81.26 0.69	68.17 🗆 2.05	72.05 🗆 🗆 2.58
HE2	297.70±16.21	86.67 0.45	74.11 🗆 🗆 1.07	76.66 🗆 🗆 2.60
HE3	335.39±21.00	88.59 0.38	78.22 🗆 🗆 1.82	80.83 🗆 🗆 3.18
C1	328.49±19.80	87.28 0.76	81.66 🗆 🗆 2.08	88.59 🗆 🗆 1.52
C2	357.21±20.26	91.53 2.02	84.00 🗆 🗆 1.26	89.70 🗆 2.02
C3	389.72±17.50	94.39 🗆 1.50	88.66 🗆 🗆 1.52	91.24 🗆 🗆 2.28
EU1	128.23±15.65	67.48 2.07	65.74 🗆 🗆 4.04	78.07 🗆 🗆 0.98
EU2	185.15±22.25	74.73 1.54	69.24 🗆 🗆 1.52	86.78 2.30
EU3	248.63 21.53	79.18 1.97	74.57 🗆 🗆 3.05	89.32 2.06

Table 2: mean particle size, Practical yield, *in vitro* buoyancy and Drug entrapment efficiencyof floating microspheres of Cephalexin







Fig. 2: In vitro release data of cephalexin loaded floating microspheres of cellulose acetate &eudragit S-100

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